

LIQUIZYME

## MAGNESIUM (XB Method)

Code	Product Name	Pack Size
LS024A	Liquizyme Magnesium	2 x 25 ml
LS024B	Liquizyme Magnesium	25 T

### Intended Use

Diagnostic reagent for quantitative *in vitro* determination of Magnesium in human serum.

### Clinical Significance

Magnesium is an essential nutrient which is involved in many biochemical functions. It has a structural role in nucleic acids and ribosomal particles, required as an activator for many enzymes and has a role in energy producing oxidative phosphorylation.

Hypomagnesaemia results in the impairment of neuromuscular functions and may develop in severe prolonged diarrhoea, malabsorption syndromes, primary aldosteronism and diuretic therapy. Hypermagnesaemia is seen in renal glomerular failure and diabetic coma.

### Principle

Magnesium reacts with Xylidyl Blue to form a colored compound in alkaline solution. The intensity of the color formed is proportional to the magnesium concentration in the specimen.

### Reagent Composition

#### Reagent 1: Magnesium Reagent

Tris Buffer : < 200 mmol/l  
Xylidyl Blue : >0.05 mmol/l

**Reagent 2 : Magnesium Standard : 2.43 mg/dl**  
Ready to use

### Reagent Preparation

Reagents are liquid. ready to use.

### Materials Required But Not Provided

- Clean & Dry container.
- Laboratory Glass Pippettes or Micropipettes & Tips.
- Colorimeter or Bio-Chemistry Analyzer.

### Stability And Storage

The unopened reagents are stable till the expiry date stated on the bottle and kit label when stored at 2–8°C.

### Specimen Collection And Handling

Use unheamolyse serum.

It is recommended to follow NCCLS procedures (or similar standardized conditions).

### Stability In Serum

7 days : at 4 – 8°C  
1 year : at -20°C

Discard contaminated specimens.



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### Calibration

Calibration with the Magnesium standard provided in the kit is recommended.

### Quality Control

It's recommended to run normal and abnormal control sera to validate reagent performance.

### Unit Conversion

mg/dl x 0.4114 = mmol/L

### Expected Values

#### Serum

Men : 1.8 - 2.6 mg/dl  
Women : 1.9 - 2.5 mg/dl  
Children : 1.5 - 2.3 mg/dl  
New Born : 1.2 - 2.6 mg/dl

**It is recommended that each laboratory verify this range or derives reference interval for the population it serves.**

### Performance Data

Limit of quantification : 0.16 mg/dl  
Linearity : 5.00 mg/dl  
Measuring range : 0.16 – 5.00 mg/dl

### Precision

Intra-assay precision Within run (n=20)	Mean (mg/dl)	SD (mg/dl)	CV (%)
Sample 1	2.15	0.03	1.23
Sample 2	3.93	0.05	1.30

Inter-assay precision Run to run (n=20)	Mean (mg/dl)	SD (mg/dl)	CV (%)
Sample 1	3.80	0.070	1.83

### Comparison

A comparison between Beacon Magnesium (y) and a commercially available test (x) using 20 samples gave following results:

y = 0.982 x + 0.005 mg/dl  
r = 0.992

### Interferences

Following substances do not interfere:

Bilirubin up to 40 mg/dl, triglycerides up to 2000 mg/dl. Haemoglobin interferes because magnesium is released by erythrocytes.

#### Warning And Precautions

For *in vitro* diagnostic use. To be handled by entitled and professionally educated person.

#### Waste Management

Please refer to local legal requirements.

#### Assay Procedure

**Wavelength** : 546 nm

**Cuvette** : 1 cm

Addition Sequence	Reagent Blank	Standard	Sample
Reagent 1	1000 µl	1000 µl	1000 µl
Standard	-	10 µl	-
Sample	-	-	10 µl
Distilled Water	10 µl	-	-

Mix and incubate 5 min. at 37°C. Measure absorbance of standard (Abs. S) and test (Abs. T) against reagent blank at 546 nm.

#### Calculation

$$\text{Magnesium (mg/dl)} = \frac{\text{Abs. T}}{\text{Abs. S}} \times 2.43$$

**Applications for automatic analysers are available on request.**

#### Assay Parameters For Photometers

Mode	End point
Wavelength 1 (nm)	546
Sample Volume (µl)	10
Reagent Volume (µl)	1000
Incubation time (min.)	5
Incubation temp. (°C)	37
Normal Low (mg/dl)	1.8
Normal High (mg/dl)	2.6
Linearity Low (mg/dl)	0.16
Linearity High (mg/dl)	5
Standard Concentration	2.43 mg/dl
Blank with	Reagent
Unit	mg/dl

#### References

1. Thomas L. Clinical Laboratory Diagnostics. 1st ed. Frankfurt: TH-Books Verlagsgesellschaft; 1998. p.231-41.
2. Endres DB, Rude RK. Mineral and bone metabolism. In: Burtis CA, Ashwood ER, editors. Tietz Textbook of Clinical Chemistry. 3rd ed. Philadelphia: W.B Saunders Company; 1999. p. 1395-1457.
3. Mann CK, Yoe JH. Spectrophotometric determination of magnesium with 1-Azo-2-hydroxy-3-(2.4-dimethyl-

carboxanilido)-naphthalene-1'-(2-hydroxybenzene). Anal Chim Acta 1957;16:155-60.

4. Bohoun C. Microdosage du magnésium dans divers milieux biologiques. Clln Chim Acta 1962;7:811-7.
5. Sitzmann FC. Normalwerte. Munchen: Hans Marseille Verlag GmbH; 1986. p.166.
6. Tietz Textbook of Clinical Chemistry and Molecular Diagnostics. Burtis, C.A., Ashwood, E.R., Bruns, D.D.; 5th edition, WB Saunders Comp., 2012

#### Symbols Used On Labels



Catalogue Number



Manufacturer



See Instruction for Use



Lot Number



Content



Storage Temperature



Expiry Date



In Vitro Diagnostics



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